

Stereoselective Preparation of Phosphine Oxides via a 2,3-Sigmatropic Shift of Allylic Diphenylphosphinites

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Abstract

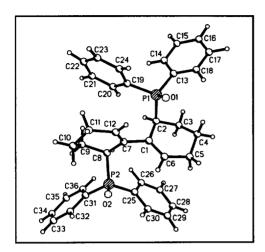
The thermic rearrangement of various chiral or racemic allylic diphenylphosphinites to allylic phosphine oxides has been applied for the preparation of several chiral diphosphine oxides of interest for asymmetric catalysis.

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The design and the preparation of chiral phosphines and diphosphines for asymmetric catalysis is an active area of research [1]. Herein, we wish to report a new approach to chiral and racemic functionalized phosphine oxides starting from readily available chiral allylic alcohols of type 1 using the [2,3] sigmatropic rearrangement of allylic diphenylphosphinites of type 2 [2] (Scheme 1). To the best of our knowledge, this rearrangement (Arbuzov rearrangement) has never been applied to chiral cyclic systems.

The 2-iodoallylic alcohols **1a-d** were prepared in a two-step sequence from the corresponding commercially available 2-cyclohexenones or 2-cyclopentenones. The enones were treated with iodine (1.1 to 1.4 equiv) in the presence of catalytic amounts of pyridinium dichromate (0.3 equiv) in dichloromethane for 6 to 24 h, affording 2-iodoenones in good yields [3]. The 2-iodoenones were reduced by the CBS method [4] with methyl-oxazaborolidine as catalyst (10-15 mol %) and BH₃.Me₂S (0.6 equiv) in THF affording the desired 2-iodoallylic alcohols **1a-d**. The enantiomeric excesses were in a range of 95% to 99% without recrystallization. The reaction of these chiral allylic alcohols with chlorodiphenylphosphine (1.05 equiv) in the presence of DMAP (1.05 equiv) in toluene furnished the intermediate phosphinites which were heated under reflux overnight. The phosphine oxides **3a**, **3b** and **3c** were isolated in good yield (Table 1). Of particular interest is the rearrangement of the phosphinite derived from the 3-substituted cyclohexenol **1c**. It furnished the phosphine oxide **3c** in 77% yield bearing a chiral quaternary center. In the case of **1d** the rearrangement failed (Scheme 1 and Table 1).

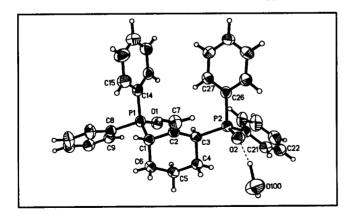
The [2,3] sigmatropic rearrangement was also performed on more elaborated systems. Thus, optically pure 1,4-diols 4a and 4b were obtained by reductive nickel catalyzed homocoupling reaction [5] from the alcohols 1a and 1b in 42% and 49% yield respectively (Zn (1 equiv), Bu₄NI (0.4 to 0.6 equiv), NiCl₂(PPh₃)₂ (0.2 to 0.3 equiv) in THF).



The double rearrangement of the corresponding biphosphinites afforded the biphosphine oxides 5a and 5b in 86% and 65% yield. In these cases, the formation of the biphosphinites had to be performed in ether and the chlorohydrate of DMAP had to be filtrated before the rearrangement. The absence of meso-5a in the crude reaction mixture confirms that these [2,3] sigmatropic rearrangements occur with complete retention of the stereochemical information. The stereochemistry of 5a was confirmed by X-ray structure (Figure 1) [8].

Figure 1. X-ray Structure of 5a

By using an unsaturated racemic 1,3-diol like 8 [6], two biphosphine oxides 10a and 10b are expected. The reaction of 8 with ClPPh₂ (2.1 equiv) and DMAP (2.1 equiv) in ether (20°C, 1h) provided the bis-phosphinite 9 which after filtration and heating under reflux in toluene for 16h underwent a double [2,3] sigmatropic rearrangement. It afforded the products 10a and 10b in a 88:12 ratio. After purification by flash chromatography, 10a and 10b were isolated in 57% and 7% yield (Scheme 2).



The product-ratio between 10a and 10b indicates that the exocyclic rearrangement is favored over the endocyclic rearrangement. The obtained X-ray structure of 10b confirms the trans-conformation between the diphenylphosphinoyl groups (Figure 2) [8]. No meso-10b was detectable by ³¹P NMR on the crude reaction mixture.

Figure 2. X-ray Structure of 10b

Table 1. Chiral phosphine oxides or diphosphine oxides obtained by a [2,3]-sigmatropic rearrangement of
phosphinites or diphosphinites prepared from enantiomerically enriched allylic alcohols 1a-d or bis-allylic
diols 4a-b.

Entry	Allylic alcohols of type 1 or 4	%ee a	[α] _D ²⁰ b	Product of type 3 or 5	Yield (%)d	[α] _D ²⁰ b
1	о́н	95 (R)	+ 60.6 (1.06)	Q: PPh2	86 (S)	- 99.3 (1.42)
2	0H	96 (R)	+ 31.9 (1.37)	Q PPh2	77 (S)	- 25.2 (1.30)
3	OH Me	99 (R)	+ 84.0 (1.86)	Ph ₂ Me	80 (S)	+ 52.0 (1.21)
4	OH I	95 (<i>R</i>)	+35.1 (1.81)	3e decomposition	-	-
5	OH OH	100 (R, R)	+ 243.0 (1.07)	PPh ₂	86 (S, S)	-143.1 (1.03)
6	41 0H	100 (R, R)	- 65.4 (1.01) ^c	5a Q PPh ₂ PPh ₂	65 (S, S)	- 69.1 (1.33)
	4b			5b		

^aThe enantiomeric excesses were determined by chiral HPLC (Chiralcel OD or OJ columns). ^bOptical rotation; the concentrations of 1a-d, 3a-c, 4a and 5a-b in CHCl₃ are given in parenthesis. ^cIn MeOH. ^dIsolated yields of analytically pure products.

Similarly, the optically pure unsaturated 1,2-diol 11 ((IR, 2R) trans cyclohex-3-en-1,2-diol) [7] was converted to the corresponding biphosphinite 12. After filtration on silicagel, heating of 12 in toluene for 42 h at reflux furnished the chiral trans 1,2-diphosphine oxide 14 ([α]_D-68.7 (c 1.23, CHCl₃)) in 74% yield via the intermediate 13 (Scheme 3).

In summary, we have described a stereoselective preparation of monophosphine oxides, 1,2-bis-phosphine oxides, 1,3-bis-phosphine oxides and 1,4-bis-phosphine oxides starting from readily available chiral or racemic allylic alcohols. The preparation of new ligands from these phosphine oxides and their application in asymmetric catalysis is currently being investigated [9].

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- [7] The preparation of the optically pure diol 11 will be reported in further article. The enantiomeric excess was determined by chiral GC analysis of the corresponding diacetate on cyclodextrin column. [α]_D -22.3 (c 1.21, CHCl₃). For the preparation of the racemic diol see: Posternak, T.; Friedli, H. Helv. Chim. Acta 1953, 36, 251-256.
- [8] X-Rays analysis were performed by measuring intensity data on a Enraf-Nonius CAD-4 diffractometer using graphite-monochromated Cu K α radiation and the ω scan technique up to $\theta=65^{\circ}$ for compound 5a and $\theta=70^{\circ}$ for compound 10b. Lists of atomic coordinates, bond distances, bond angles, torsional angles have been deposited at the Cambridge Crystallographic Data Center (12 Union Road, Cambridge CB2 1EZ, UK). Deposition numbers of 5a and 10b are CCDC 118281 and CCDC 118282.

[9] Typical Procedures

Method A: A 250-mL-flask under argon equipped with a magnetic stirrer and a reflux condenser is charged with (R)-2-iodo cyclohexenol (3.36 g, 15 mmol), DMAP (1.93 g, 15.8 mmol, 1.05 eq) and toluene (80 mL freshly distilled over sodium). To this homogeneous solution is added dropwise pure chlorodiphenylphosphine (2.85 mL, 3.49 g, 15.8 mmol, 1.05 eq). The resulting suspension is stirred at rt for 1 h and then heated to reflux for 16 h. The mixture is cooled to rt, diluted with CH₂Cl₂ and ether and washed successively with a 10 % solution of HCl, water and brine. The organic phase is dried over MgSO₄, filtrated and the solvents evaporated *in vacuo*. The crude compound is purified by chromatography over silica gel (CH₂Cl₂/Et₂O 1:1) yielding pure 3a as a colorless crystalline solid (5.25 g, 86 %; mp = 193-195 °C). The same procedure is applied for phosphine oxides 3b and 3c.

Method B: A 100-mL-flask under argon equipped with a magnetic stirrer is charged with (1R, 1'R)2, 2-bicyclohexenol (157 mg, 0.808 mmol), DMAP (218 mg, 1.78 mmol, 2.2 eq) and ether (15 mL freshly distilled over sodium). To the homogeneous solution, pure chlorodiphenylphosphine (0.31 mL, 0.375 g, 2.1 eq, 1.70 mmol) is added dropwise. The resulting suspension is stirred at rt for 1 h, filtrated under argon over Celite and washed twice with ether. The etheral solution is evaporated under reduced pressure and the resulting solid biphosphinite is diluted in toluene (30 mL freshly distilled over sodium). The resulting solution is refluxed under argon for 20 h and then cooled to rt. The work-up corresponds to the procedure A. The crude compound is purified by column chromatography (Et₂O/CH₂Cl₂/MeOH 60/40/0 to 60/40/5) yielding 5a as a colorless crystalline solid (396 mg, 86 %; mp = 270-274 °C with decomposition).